

P1.4 At time $t=0$, a particle is represented by the wavefunction

$$\Psi(x, 0) = \begin{cases} A\left(\frac{x}{a}\right), & 0 \leq x \leq a \\ A\left(\frac{b-x}{b-a}\right), & a \leq x \leq b \\ 0 & \text{otherwise} \end{cases}$$

where A , a , and b are positive constants.

- Normalize Ψ (Find A in terms of a and b) ✓
- Sketch $\Psi(x, 0)$ as a function of x ✓
- Where is the particle most likely to be found at $t=0$? ✓
- What is the probability of finding the particle to the left of a ? ✓
Check your result in the limiting cases $b=a$ and $b=2a$ ✓
- What is the expectation value of x ? ✓

Integrals over other intervals yield 0

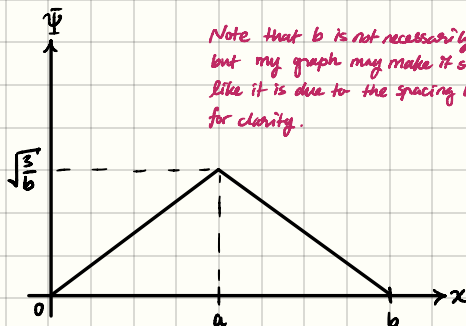
$$(a) \int_{-\infty}^{\infty} |\Psi(x, 0)|^2 dx = 1 \Rightarrow \int_0^a A^2 \frac{x^2}{a^2} dx + \int_a^b A^2 \left(\frac{b-x}{b-a}\right)^2 dx = 1$$

$$\Rightarrow \frac{A^2}{a^2} \left(\frac{x^3}{3} \right)_0^a + \frac{A^2}{(b-a)^2} \int_a^b (b^2 - 2bx + x^2) dx = \frac{1}{3} A^2 a + \frac{A^2}{(b-a)^2} \left(b^2 x - bx^2 + \frac{x^3}{3} \right)_a^b = 1$$

$$\Rightarrow A^2 \left[\frac{1}{3} a + \frac{1}{(b-a)^2} \left(b^3 - b^2 a + \frac{b^3}{3} - b^2 a + ba^2 - \frac{a^3}{3} \right) \right] = A^2 \left(\frac{1}{3} a + \frac{1}{3} b - \frac{1}{3} a \right) = 1$$

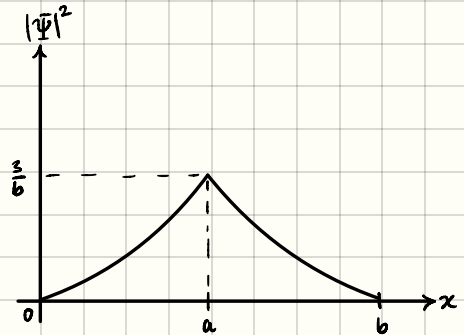
$$\Rightarrow A = \sqrt{\frac{3}{b}}$$

$$(b) \Psi(x, 0) = \begin{cases} \sqrt{\frac{3}{b}} \left(\frac{x}{a} \right) & 0 \leq x \leq a \\ \sqrt{\frac{3}{b}} \left(\frac{b-x}{b-a} \right) & a \leq x \leq b \\ 0 & \text{otherwise} \end{cases}$$



(c) We can guess from the graph for $|\Psi|^2$ below that it will be $x = a$.

$$|\Psi(x,0)|^2 = \begin{cases} \frac{3}{b} \left(\frac{x}{a}\right)^2 & 0 \leq x \leq a \\ \frac{3}{b} \left(\frac{b-x}{b-a}\right)^2 & a \leq x \leq b \\ 0 & \text{otherwise} \end{cases}$$



$$(d) P(x \leq a) = \int_0^a \frac{3}{b} \left(\frac{x}{a}\right)^2 dx = \frac{3}{a^2 b} \frac{x^3}{3} \Big|_0^a = \frac{a}{b}$$

Check for $b = a \rightarrow P(x \leq a) = 1$

Check for $b = 2a \rightarrow P(x \leq a) = \frac{1}{2}$

$$(e) \langle x \rangle = \int_{-\infty}^{\infty} x |\Psi|^2 dx = \int_0^a \frac{3}{b} \frac{x^3}{a^2} dx + \int_a^b \frac{3}{b} x \left(\frac{b-x}{b-a}\right)^2 dx$$

$$= \frac{3}{a^2 b} \frac{a^4}{4} + \frac{3}{b(b-a)^2} \int_a^b x(b-x)^2 dx \quad u = x-b, du = dx$$

$$= \frac{3}{4} \frac{a^2}{b} + \frac{3}{b(b-a)^2} \int_{x=a}^{x=b} (u+b)u^2 du = \frac{3}{4} \frac{a^2}{b} + \frac{3}{b(b-a)^2} \int_{x=a}^{x=b} u^3 + bu^2 du$$

$$= \frac{3}{4} \frac{a^2}{b} + \frac{3}{b(b-a)^2} \left(\frac{u^4}{4} + b \frac{u^3}{3} \right) \Big|_{x=a}^{x=b} = \frac{3a^2}{4b} + \frac{3}{b(b-a)^2} \cdot \left(\frac{(x-b)^4}{4} + \frac{b(x-b)^3}{3} \right) \Big|_a^b$$

$$= \frac{3a^2}{4b} + \frac{3}{b(b-a)^2} \cdot \frac{(b-a)^3 (b+3a)}{4}$$

$$= \frac{3a^2 + b^2 + 2ab - 3a^2}{4b}$$

$$= \frac{b+2a}{4}$$

P1.9 A particle of mass m has the wave function

$$\Psi(x,t) = A e^{-a \left[\frac{mx^2}{\hbar} + it \right]}$$

where A and a are positive real constants

(a) Find A . ✓

(b) For what potential energy function, $V(x)$, is this a solution to the Schrödinger equation? ✓

(c) Calculate the expectation values of x , x^2 , p , and p^2 . ✓

(d) Find σ_x and σ_p . Is their product consistent with the uncertainty principle? ✓

$$(a) \int_{-\infty}^{\infty} |\Psi(x,t)|^2 dx = 1 \Rightarrow \int_{-\infty}^{\infty} \Psi \Psi^* dx = 1$$

$$\Rightarrow \int_{-\infty}^{\infty} A^2 e^{-2a \frac{mx^2}{\hbar}} \cancel{e^{-ait}} \cancel{e^{ait}} dx = A^2 \int_{-\infty}^{\infty} e^{-\left(\frac{2am}{\hbar}\right)x^2} dx = 2A^2 \int_0^{\infty} e^{-\frac{x^2}{\frac{\hbar}{2am}}} dx = 1$$

$n=0 \rightarrow \sqrt{\pi} \frac{2(0)!}{(0)!} \left(\frac{a}{2}\right)^{2(0)+1}$

$$\Rightarrow 2A^2 \sqrt{\pi} \cdot \sqrt{\frac{\hbar}{2am}} \cdot \frac{1}{2} = A^2 \sqrt{\frac{\pi \hbar}{2am}} = 1 \Rightarrow A = \left(\frac{2am}{\pi \hbar} \right)^{1/4}$$

$$(b) i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V\Psi$$

$$\frac{\partial \Psi}{\partial t} = -aiAe^{-a \left[\frac{mx^2}{\hbar} + it \right]}$$

$$\frac{\partial^2 \Psi}{\partial x^2} = \frac{\partial}{\partial x} \left(-\frac{2amx}{\hbar} A e^{-a \left[\frac{mx^2}{\hbar} + it \right]} \right) = -\frac{2am}{\hbar} A e^{-a \left[\frac{mx^2}{\hbar} + it \right]} + \frac{4a^2 m^2 x^2}{\hbar^2} A e^{-a \left[\frac{mx^2}{\hbar} + it \right]}$$

$$V\Psi = i\hbar \left(-aiAe^{-a \left[\frac{mx^2}{\hbar} + it \right]} \right) - \frac{\hbar^2}{2m} A e^{-a \left[\frac{mx^2}{\hbar} + it \right]} \left(\frac{2am}{\hbar} - \frac{4a^2 m^2 x^2}{\hbar^2} \right)$$

$$V(x) = a\hbar - \frac{\hbar^2}{2m} \left(\frac{2am}{\hbar} - \frac{4a^2 m^2 x^2}{\hbar^2} \right) = a\hbar - a\hbar + 2a^2 m x^2 = 2a^2 m x^2$$

$$(c) \langle x \rangle = \int_{-\infty}^{\infty} x |\Psi|^2 dx = A^2 \int_{-\infty}^{\infty} x e^{-\left(\frac{2am}{\hbar}\right)x^2} dx = 0$$

$$\overset{f(x)}{f(-x)} = -x e^{-\frac{2am}{\hbar}x^2} = -f(x) \Rightarrow \text{odd}$$

$$\langle x^2 \rangle = 2A^2 \int_0^{\infty} x^2 e^{-\left(\frac{2am}{\hbar}\right)x^2} dx = 2A^2 \sqrt{\pi} \frac{[2(1)]!}{(1)!} \left(\frac{\alpha}{2}\right)^3 = \frac{1}{2} \sqrt{\frac{2am}{\hbar}} \sqrt{\pi} \cdot \left(\frac{\hbar}{2am}\right)^{3/2}$$

$$= \frac{\hbar}{4am}$$

$$\langle p \rangle = -i\hbar \int \Psi^* \frac{\partial \Psi}{\partial x} dx = -i\hbar \int_{-\infty}^{\infty} A e^{-a\left(\frac{mx^2}{\hbar} - it\right)} \cdot \left(-\frac{2amx}{\hbar} A e^{-a\left(\frac{mx^2}{\hbar} + it\right)}\right) dx$$

$$= +i\hbar A^2 \cdot \frac{2am}{\hbar} \underbrace{\int_{-\infty}^{\infty} x e^{-2a\frac{mx^2}{\hbar}} dx}_{\text{odd}} = 0$$

$$\langle p^2 \rangle = -\hbar^2 \int \Psi^* \frac{\partial^2 \Psi}{\partial x^2} dx = +\hbar^2 \int_{-\infty}^{\infty} A e^{-a\left(\frac{mx^2}{\hbar} - it\right)} \cdot A e^{-a\left[\frac{mx^2}{\hbar} + it\right]} \left(\frac{2am}{\hbar} - \frac{4a^2 m^2 x^2}{\hbar^2}\right) dx$$

$$= 2\hbar^2 A^2 \int_0^{\infty} e^{-\frac{2amx^2}{\hbar}} \cdot \frac{2am}{\hbar} \left(1 - \frac{2amx^2}{\hbar}\right) dx$$

$$= 2\hbar^2 \sqrt{\frac{2am}{\hbar}} \cdot \frac{2am}{\hbar} \left[\frac{\sqrt{\pi}}{2} \left(\frac{\hbar}{2am}\right)^{3/2} - \left(\frac{2am}{\hbar}\right)^{3/2} \frac{\sqrt{\pi}}{4} \cdot \frac{1}{8} \left(\frac{\hbar}{2am}\right)^{3/2} \right]$$

$$= \hbar^2 \frac{4am}{\hbar} \sqrt{\frac{2am}{\hbar}} \left[\frac{1}{2} \left(\frac{\hbar}{2am}\right)^{3/2} - \frac{1}{4} \left(\frac{\hbar}{2am}\right)^{3/2} \right]$$

$$= \hbar^2 \frac{4am}{\hbar} \cdot \frac{1}{4} = \hbar am$$

$$(d) \sigma_x = \sqrt{\langle x^2 \rangle - \langle x \rangle^2} = \sqrt{\frac{\hbar}{4am}}$$

$$\sigma_p = \sqrt{\langle p^2 \rangle - \langle p \rangle^2} = \sqrt{\hbar am}$$

$$\sigma_x \sigma_p = \sqrt{\frac{\hbar^2 am}{4am}} = \frac{\hbar}{2} \geq \frac{\hbar}{2} \Rightarrow \text{consistent with Heisenberg uncertainty principle!}$$

Pl.14 Let $P_{ab}(t)$ be the probability of finding the particle in the range $(a < x < b)$, at time t .

(a) Show that $\frac{dP_{ab}}{dt} = J(a, t) - J(b, t)$, switch? ✓

where $J(x, t) \equiv \frac{i\hbar}{2m} \left(\Psi \frac{\partial \Psi^*}{\partial x} - \Psi^* \frac{\partial \Psi}{\partial x} \right)$. ("probability current")

+ What are the units of $J(x, t)$? ✓

(b) Find the probability current for the wave function in Pl.9. ✓

(a) $P_{ab}(t) = \int_a^b |\Psi(x, t)|^2 dx = \int_a^b \Psi^*(x, t) \Psi(x, t) dx$

$\frac{dP_{ab}}{dt} = \frac{d}{dt} \int_a^b \Psi^*(x, t) \Psi(x, t) dx = \int_a^b \frac{\partial}{\partial t} \Psi^*(x, t) \Psi(x, t) dx$

$= \int_a^b \frac{\partial \Psi^*}{\partial t} \Psi + \Psi^* \frac{\partial \Psi}{\partial t} dx$

$\left\{ \begin{array}{l} \frac{\partial \Psi}{\partial t} = \frac{i}{\hbar} \frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + \frac{i}{\hbar} V \Psi \\ \frac{\partial \Psi^*}{\partial t} = -\frac{i}{\hbar} \frac{\hbar^2}{2m} \frac{\partial^2 \Psi^*}{\partial x^2} - \frac{i}{\hbar} V \Psi^* \end{array} \right.$

(Schrödinger Equation)

$= \frac{i\hbar}{2m} \int_a^b \left(-\frac{\partial^2 \Psi^*}{\partial x^2} \Psi - \cancel{\frac{2m}{\hbar^2} V \Psi^* \Psi} + \Psi^* \frac{\partial^2 \Psi}{\partial x^2} + \cancel{\frac{2m}{\hbar^2} \Psi^* V \Psi} \right) dx$

$= \frac{i\hbar}{2m} \int_a^b \left(\Psi^* \frac{\partial^2 \Psi}{\partial x^2} - \Psi \frac{\partial^2 \Psi^*}{\partial x^2} \right) dx$

$= \frac{i\hbar}{2m} \int_a^b \left(\underbrace{\Psi^* \frac{\partial^2 \Psi}{\partial x^2}}_{\frac{\partial}{\partial x} \left(\Psi^* \frac{\partial \Psi}{\partial x} \right)} - \underbrace{\Psi \frac{\partial^2 \Psi^*}{\partial x^2}}_{-\frac{\partial}{\partial x} \left(\Psi \frac{\partial \Psi^*}{\partial x} \right)} + \frac{\partial \Psi^*}{\partial x} \frac{\partial \Psi}{\partial x} - \frac{\partial \Psi^*}{\partial x} \frac{\partial \Psi}{\partial x} \right) dx$

$= \frac{i\hbar}{2m} \int_a^b \frac{\partial}{\partial x} \left(\Psi^* \frac{\partial \Psi}{\partial x} - \Psi \frac{\partial \Psi^*}{\partial x} \right) dx$

Fundamental
Theorem of
Calculus

$= \frac{i\hbar}{2m} \left(\Psi^*(b, t) \frac{\partial \Psi(b, t)}{\partial x} - \Psi(b, t) \frac{\partial \Psi^*(b, t)}{\partial x} \right) - \frac{i\hbar}{2m} \left(\Psi^*(a, t) \frac{\partial \Psi(a, t)}{\partial x} - \Psi(a, t) \frac{\partial \Psi^*(a, t)}{\partial x} \right)$

$= J(b, t) - J(a, t)$

$[s^{-1}]$

from $\frac{dP_{ab}}{dt} \leftarrow \text{unitless}$
 $\frac{d}{dt} \leftarrow [s]$

$$(b) \Psi(x,t) = A e^{-a \left[\frac{mx^2}{\hbar} + it \right]}$$

$$\Psi^*(x,t) = A e^{-a \left[\frac{mx^2}{\hbar} - it \right]}$$

$$\frac{\partial \Psi}{\partial x} = -\frac{2amx}{\hbar} A e^{-a \left[\frac{mx^2}{\hbar} + it \right]}$$

$$\frac{\partial \Psi^*}{\partial x} = -\frac{2amx}{\hbar} A e^{-a \left[\frac{mx^2}{\hbar} - it \right]}$$

$$J(x,t) = \frac{i\hbar}{2m} \left(A e^{-a \left[\frac{mx^2}{\hbar} - it \right]} \cdot -\frac{2amx}{\hbar} A e^{-a \left[\frac{mx^2}{\hbar} + it \right]} + A e^{-a \left[\frac{mx^2}{\hbar} + it \right]} \cdot \frac{2amx}{\hbar} A e^{-a \left[\frac{mx^2}{\hbar} - it \right]} \right)$$

$$= 0$$

P1.18

Very roughly speaking, quantum mechanics is relevant when the de Broglie wavelength of the particle in question ($\frac{h}{p}$) is greater than the characteristic size of the system (d). In thermal equilibrium at (Kelvin) temperature T , the average kinetic energy of a particle is

$$\frac{p^2}{2m} = \frac{3}{2} k_B T \quad (\text{where } k_B \text{ is Boltzmann's constant}), \text{ so the typical}$$

$$\text{de Broglie wavelength is } \lambda = \frac{h}{\sqrt{3mk_B T}}$$

The purpose of this problem is to determine which systems will have to be treated quantum mechanically, and which can safely be described classically.

(a) **Solids.** The lattice spacing in a typical solid is around $d = 0.3 \text{ nm}$. Find the temperature below which the unbound electrons in a solid are quantum mechanical. ✓

+ Below what temperature are the nuclei in a solid quantum mechanical? ✓ (Use silicon as an example.)

Moral: The free electrons in a solid are always quantum mechanical; the nuclei are generally not quantum mechanical.

(b) **Gases.** For what temperatures are the atoms in an ideal gas at pressure P quantum mechanical? Hint: Use the ideal gas law ($PV = Nk_B T$) to deduce all the interatomic spacing. ✓

$$\text{Answer: } T < \frac{1}{k_B} \left(\frac{h^2}{3m} \right)^{3/5} P^{2/5}. \text{ Obviously (for the gas to show quantum}$$

behaviour) we want m to be as small as possible, and P as large as possible. Put in the numbers for Helium at atmospheric pressure. ✓ Is hydrogen in outer space (where the interatomic spacing is about 1 cm and the temperature is at least 3 K) quantum mechanical? ✓ (Assume it's monatomic hydrogen, not H_2 .)

$$(a) \quad \frac{h}{p} > d \rightarrow \lambda > d \rightarrow \frac{h}{\sqrt{3mk_B T}} > d$$

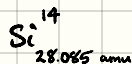
$$\Rightarrow T < \frac{h^2}{3mk_B d^2}$$

① electrons: $m_e = 9.109 \times 10^{-31} \text{ kg}$ $h = 6.626 \times 10^{-34} \text{ J}\cdot\text{s}$
 $k_B = 1.381 \times 10^{-23} \text{ J}\cdot\text{K}^{-1}$ $d = 0.3 \times 10^{-9} \text{ m}$

$$T < 1.293 \times 10^5 \text{ K}$$

② Si nuclei: $m = 28.085 \text{ amu} \times \frac{1.661 \times 10^{-27} \text{ kg}}{1 \text{ amu}} = 4.665 \times 10^{-26} \text{ kg}$

mass of Si nucleus



$1 \text{ amu} = 1.661 \times 10^{-27} \text{ kg}$

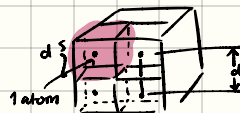
with same constants as before and same assumed lattice spacing $d = 0.3 \text{ nm}$:

$$T < 2.525 \text{ K}$$

(b) $PV = Nk_B T \Rightarrow Pd^3 = Nk_B T \rightarrow d = \left(\frac{k_B T}{P} \right)^{1/3}$

$$\frac{h}{\sqrt{3mk_B T}} > \left(\frac{k_B T}{P} \right)^{1/3} \rightarrow \frac{h^3 P}{(3m)^{3/2} k_B^{5/2}} > T^{5/2}$$

$$\Rightarrow T < \frac{P^{2/5} h^{6/5}}{(3m)^{3/5} k_B}$$



For helium:

$m = 4.003 \text{ amu} \times \frac{1.661 \times 10^{-27} \text{ kg}}{1 \text{ amu}} = 6.649 \times 10^{-27} \text{ kg}$

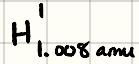


$1 \text{ atm} = 1.013 \times 10^5 \text{ Pa}$

$P = 1.013 \times 10^5 \text{ Pa}$

$$T < 2.936 \text{ K}$$

Far monatomic hydrogen:



$$m = 1.008 \text{ amu} \times \frac{1.661 \times 10^{-27} \text{ kg}}{1 \text{ amu}} = 1.674 \times 10^{-27} \text{ kg}$$

$$d = 0.01 \text{ m}$$

$$T < \frac{h^2}{3mk_B d^2} \Rightarrow T < 6.332 \times 10^{-14} \text{ K} \Rightarrow \text{MUCH smaller than 3K}$$

$\Rightarrow$ No, H in outer space is NOT quantum mechanical